SPLITTING OF THE DEGENERATE ACETYLENIC **FTMOS**; A probe for ring strain ¹

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As Figure 1 shows, the splitting between the π MOs of acetylene depends strongly on the angle α , i.e. the deviation from the colinear arrangement of the C-H bonds and the triple bond. The destabilization of the π_y MO is mainly due to the negative overlap ($S_{1.205} = -0.29$) between the $2p_x$ AOs contributing to this MO for $\alpha \neq 0$ ($0 < \alpha < 90^{\circ}$). Recently synthesized systems, namely cyclooctyne ($\underline{1}$)², 3,3,8,8-tetramethylcyclooctyne ($\underline{2}$)³, 3,3,7,7-tetramethylcycloheptyne ($\underline{3}$)⁴, and 3,3,6,6-tetramethyl-1-thiacycloheptyne ($\underline{4}$)⁵ allow an experimental examination of these theoretical predicti-



ons.

Figure 2 shows the photoelectron (pe) spectra of <u>3</u> and <u>4</u> and Figure 3 displays the experimental⁶ correlation diagram for the highest

Figure 1

Calculated (MINDO/2 method¹²) splitting of the π_y and π_z MOs in acetylene versus the angle \ll .

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Figure 2

Photoelectron spectra 13 of molecules 3 and 4.

occupied MOs (the π_v and π_z MOs of the triple bond and the n- lone pair MO on the sulphur atom of 1 to 4 and of 3,3,6,6-tetramethyl-1-thiacycloheptane(5). As the spectra and correlation diagram (Figure 3) reveal, the degenerate acetylenic π MOs are not split in either 1 or 2. For $1, \propto$ is 21.5°⁷. The expected splitting (Figure 1) is only about 0.2 eV and thus not resolved in the pe spectrum of 1⁸. In the case of 3, α amounts to 34.2°⁹. In agreement with the predicted splitting of about 0.5 eV, the corresponding bands in the pe spectra (Figure 2) of 3 and 4 are split (measured splitting = 0.32 eV for 3 and 0.31 eV for 4). These results show that with increasing o(, i.e. increasing ring strain, the splitting of the degenerate acetylenic TTMOs grows and can be observed, for the first time, with the cycloheptyne systems.



Figure 3

Correlation diagram for the highest occupied MOs of molecules $\underline{1}$ to $\underline{5}$.



Figure 4

CNDO/S ¹¹ calculations on $\underline{4}$ at three stages [(1) to (3)] of conjugative interruption.

From our results we may additionally expect that for cyclohexyne systems (e.g. the benzyne system <u>6</u>) the splitting of the acetylenic π MOs should be as high as about 1 eV. Moreover, our findings suggest that the increase in the π_y <u>MO energy</u> in strained cycloalkynes is the source of the enhanced reactivity of the triple bond in these systems.

In order to confirm that the observed splitting in $\frac{4}{2}$ is due to ring strain and not to possible transannular interaction between

the $n_{\mathcal{T}}$ and the \mathcal{T}_{y} MOs we used a method ¹⁰ which permits one to interrupt the conjugation of a selected portion of molecule with the rest of the system in conjunction with an extended CNDO/S method ¹¹. The results are shown in Figure 4. Obviously, the splitting in case (2) is only insignificantly higher than in case (1), indicating that the transannular $n_{\mathcal{T}}$ / \mathcal{T}_{y} interaction plays no rôle.



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